**Original** Article

# Newton's Method Cubic Equation of State C++ Source Code for Iterative Volume Computation

Abdulhalim Musa Abubakar<sup>1</sup>, Ahmad Abubakar Mustapha<sup>2</sup>

<sup>1</sup>Department of Chemical Engineering, Modibbo Adama University of Science and Technology (MAUTECH), Girei LGA, Yola, Nigeria

<sup>2</sup>Department of Computer Science & Engineering, Vigan's Foundation for Science, Technology & Research (VFSTR) (Deemed to be University), Vadlamudi, Guntur, A.P., India

Received: 25 April 2021

Revised: 26 May 2021

Accepted: 11 June 2021

Published: 24 June 2021

**Abstract** - This write-up enumerates the basic cubic equations of state used for pressure, volume and temperature determination for pure substances. It, however, focuses on developing a Dev C++ program for molar volume computation. Since molar volume in all cubic equations of state appears unsolvable due to their nonlinear nature, numerical solution methods, one of which is the Newton-Raphson method, can be utilized to estimate the root of such polynomial equations. The 257 lines of code running from Figure 1 to Figure 6, if executed, will return the molar volume to some certain iteration over a tolerance error specified. The program is a 4-in-1 calculator. The user is capable of working with a desired equation of state at time. Users should note that the source code converges for either large or small errors specified.

**Keywords** - C++ program, Cubic equations of state, Equation of state, Newton's method, Iteration.

# **1. Introduction**

An Equation of State (EOS) is a semi-empirical functional relationship between the pressure (P), volume (V) and temperature (T) of a pure substance (Sahay, Edison and Mohamed). The simplest EOS is the ideal gas law itself. As per EOS, every new equation is theoretically believed to be superior to those that precede it. One equation is best for calculating density but not for vapor pressure, while another is very accurate for predicting vapor pressure but not for density. There are different types of EOS that fall into three categories (Mansour): a) First-class EOS are basically cubic equations of state. The cubic equations of state such as the Van der Waals, Redlich and Kwong, Soave-Redlich-Kwong, and Peng-Robinson equations give reasonable results for the thermodynamic behavior of real fluids; b) Second class EOS are non-cubic. They provide accurate results for both vapor and liquid phases, of which the Benedict et al. equation is a good example for this class of equations; and c) Third class EOS which are non-analytical EOS that are highly constrained for some specific fluids. These equations of states, especially those requiring iteration to determine a given parameter, can be difficult to handle. Numerical analysis can be applied here, and one such method is Newton's Method. The method starts by taking an initial value or guesses to get a new approximation to the root. These other new values would then serve as initial to get the next and so on. However, doing this manually might take

time, especially if it is not converging quickly. A program can be written to forever prevent these grueling molar volume calculations. An example of such a programming languages is C++.

The program did not, however, include a write-up that evaluates P, T and the number of moles (n) for the cubic EOS, as these parameters don't require iterations. Where molar volume is substituted with (V/n), the cubic EOS, an equation showing the number of moles (n) with up to degree 3 is seen. Such an equation, too, requires iteration. The temperature under Redlich-Kwong EOS would as well require iterations. Deiters et al. (2014) wrote in "Calculation of Densities from Cubic Equations of State: Revisited" but did not, however, capture the programming aspect of finding the densities or molar volume of the EOS.

# 2. Comparison of cubic EOS

Cubic equations of state (CEOS) are widely used in phase-equilibrium calculations because of their simplicity and accuracy (Soedarto). The cubic equations of state (CEOS) such as Van der Waals, Redlich-Kwong, Soave, and Peng-Robinson are simple models that have been widely used in the oil industry and also to describe the state of reservoir fluids at given conditions (Mansour). Table 1 displays the four CEOS and their constant parameter expressions (Himmelblau and Riggs): Table 1 Cubic constiant of state

Table 1. Cubic equations of state				
Van der Waals EOS	Redlich-Kwong EOS			
$\left(P + \frac{a}{\hat{v}^2}\right)\left(\hat{V} - b\right) = RT$ where $a = \frac{27}{64} \frac{R^2 T_c^2}{P_c}$ and $b = \frac{RT_c}{8P_c}$	$\begin{bmatrix} P + \frac{a}{T^{\frac{1}{2}}\hat{V}(\hat{V} + b)} \end{bmatrix} (\hat{V} - b) = RT$ where $a = 0.42748 \frac{R^2 T_c^{2.5}}{P_c}$ and			
	$b = 0.08664 \frac{RT_c}{P_c}$			
Soave Redlich-Kwong EOS	Peng-Robinson EOS			
$P = \frac{RT}{\hat{V} - b} - \frac{a \alpha}{\hat{V}(\hat{V} + b)}$	$P = \frac{RT}{\hat{V} - b} - \frac{a \alpha}{\hat{V}(\hat{V} + b) + b(\hat{V} - b)}$			
where $a = 0.42747 \frac{R^2 T_c^2}{P_c}$ ;	where $a = 0.45724 \frac{R^2 T_c^2}{P_c}$ and			
$b = 0.08664 \frac{RT_c}{P_c} \text{ and } T_r = \frac{T}{T_c}$	$b = 0.07780 \frac{RT_c}{P_c}$			
$b = 0.08664 \frac{RT_c}{P_c} \text{ and } T_r = \frac{T}{T_c}$ $\alpha = \left[1 + (0.48508 + 1.55171w - 0.15613w^2)(1 - T_r^{0.5})\right]^2$	$b = 0.07780 \frac{RT_c}{P_c}$ $\alpha = \left[1 + k\left(1 - T_r^{\frac{1}{2}}\right)\right]^2  \text{where}  T_r = \frac{T}{T_c}$ $k = 0.37464 + 1.54226w - 0.26992w^2$			

The van der Waals EOS is the basis of various EOS. It is not capable of predicting phase equilibrium accurately. Redlich-Kwong EOS needs critical temperature and pressure for P, V or T calculations. This equation is not accurate enough for density and phase-equilibria calculations, including vapor pressures and solubility of solids in supercritical fluids (Soedarto).

The Redlich Kwong EOS can satisfactorily be applied for fugacity, enthalpy and entropy departure calculations as well as gas phase properties. This equation poorly predicts liquid phase properties (Sahay, Edison and Mohamed). The Soave-Redlich-Kwong (SRK) and Peng Robinson EOS are the most applicable in the petroleum industry (Oliveira, Ribeiro and Queimada). Soave replaced the term  $T^{-0.5}$  of the RK equation of state by a function  $\alpha$  involving temperature and acentric factor. The parameter was formulated to make the equation fit the vapor pressure data of hydrocarbons (Housam and Zakia).

The Peng-Robinson EOS was suggested to satisfy the following objectives (Housam and Nasri, Applications of the peng-robinson equation of state using MATLAB): a) parameters of this EOS should be defined in terms of the critical properties and the acentric factor; b) reasonable accuracy near the critical point, particularly for calculations of the compressibility factor and liquid density; c) a single binary interaction parameter, which should be independent of temperature, pressure and composition, is needed for the

mixing rules; and d) PENG EOS should be applicable in natural gas processes.

The PENG EOS provided results similar to the SRK EOS, although it is generally superior in estimating the liquid densities of many materials, especially nonpolar ones. The applicability of the Peng Robinson equation of state in the computation of thermodynamic interactions of volatile organic compounds and biodiesel has also been tested by Sahay et al. (2013).

Working with this set of equations might sometimes be a tedious task, especially if the user is after the molar volume  $(\hat{V})$  given other conditions. The term "cubic" in CEOS earns its name because they result in a polynomial of degree 3 in volume if manipulated. These types of nonlinear equations can be solved by applying Numerical Methods like Secant, Fixed-point iteration, Regula-Falsi, Bisection or Newton-Raphson Method.

## 3. Application of Newton-Raphson Method

This Journal hand-picked the Newton-Raphson analytical method because of its rapid convergence to the root as one reason. By Newton's Method, if *f* is differentiable and *x* is an approximate solution of equation f(x) = 0, then a better approximation might be obtained using  $x_{i+1} = x_i - \frac{f(x_i)}{f'(x_i)}$ , for i = 0, 1, 2, ..., where i = number of iterations or repetitions (Mate).

EOS in Table 1 can be rewritten in the following way to compute molar volume  $(\hat{V})$  of a given gas:

## 3.1. Van der Waals EOS

Writing the Van der Waals equation as a function of molar volume:

$$f(\hat{V}) = \hat{V}^3 - \left(b + \frac{RT}{P}\right)\hat{V}^2 + \left(\frac{a}{P}\right)\hat{V} - \frac{ab}{P} = 0$$
(1)

$$f'(\hat{V}) = 3\hat{V}^2 - 2\left(b + \frac{RT}{P}\right)\hat{V} + \frac{a}{P}$$
 (2)

#### 3.2. Redlich-Kwong EOS

The Redlich-Kwong EOS written as function of volume,  $f(\hat{V})$  is as follows

$$f(\hat{V}) = \hat{V}^3 - \left(\frac{RT}{p}\right)\hat{V}^2 - \left(b^2 + \frac{RTb}{p} - \frac{a}{pT^{\frac{1}{2}}}\right)\hat{V} - \frac{ab}{pT^{\frac{1}{2}}} = 0 \quad (3)$$

$$f'(\hat{V}) = 3\hat{V}^2 - \left(\frac{2RT}{P}\right)\hat{V} - \left(b^2 + \frac{RTb}{P} - \frac{a}{PT^{\frac{1}{2}}}\right) \quad (4)$$

### 3.3. Soave Redlich-Kwong EOS

SRK EOS can be presented in the following form:

$$f(\hat{V}) = \hat{V}^3 - \left(\frac{RT}{P}\right)\hat{V}^2 - \left(b^2 + \frac{RTb}{P} - \frac{a\,\alpha}{P}\right)\hat{V} - \frac{ab\alpha}{P} = 0 \quad (5)$$

$$f'(\hat{V}) = 3\hat{V}^2 - \left(\frac{2RT}{P}\right)\hat{V} - \left(b^2 + \frac{RTb}{P} - \frac{a\,\alpha}{P}\right) \quad (6)$$

## 3.4. Peng-Robinson

**Rewriting Peng-Robinson EOS:** 

$$f(\hat{V}) = \hat{V}^3 + \left(b - \frac{RT}{p}\right)\hat{V}^2 - \left(3b^2 + \frac{2RTb}{p} - \frac{a\,\alpha}{p}\right)\hat{V} + \frac{b^3 + RTb^2 - ab\alpha}{p} = 0$$
(7)

$$f'(\hat{V}) = 3\hat{V}^2 + 2\left(b - \frac{RT}{P}\right)\hat{V} - \left(3b^2 + \frac{2RTb}{P} - \frac{a\,\alpha}{P}\right) \tag{8}$$

For all the EOS, we have:

$$\hat{V}_{i+1} = \hat{V}_i - \frac{f(\hat{V}_i)}{f'(\hat{V}_i)}$$
 (9)

There is always a starting value (say  $\hat{V}_0$ ) to help calculate the root of equations (1), (3), (5) and (7). For a given problem, guess the volume calculated from ideal gas law as the initial value (that is  $\hat{V}_{ideal} = \frac{RT}{p}$ ). Going the manual way by paper and calculator is quite time-consuming and makes one prone to error. In some situations, available online web calculators for any of the EOS might help, but one might have to grapple with unit conversion as there is hardly an online web calculator that makes provision for selecting different units.

The easiest or fastest of all is the use of MS Excel. Limitation common to all the aforementioned solution approaches is that one has to face the same stress for every new problem encountered/given. Therefore, this write-up wishes to present a multiple-option selection C++ program that computes molar volume,  $\hat{V}$ , by Newton's iterations for the CEOS.

# 4. C++ Source Code for Molar Volume Evaluation

The C++ language is a better version of the C language. One of its main features is that it supports object-oriented programming (OOP). C++ is used for scientific computing as well as system programming (Ghosh).

The below C++ syntax uses a do-while statement to solve for numerical solutions or roots of nonlinear equations by Newton's method:

CEOS Volume Iteration.cpp

```
/* Iterative Volume Computation by Newton's Method for Cubic EOS */
 1
 2
      #include <iostream>
 3
     #include <math.h>
     #include <iomanip>
 4
 5
     using namespace std;
 6
     // VANDER WAALS EOS
 7
     #define f(V) V*V*V-SE1*V*V+(aVDW/P)*V-(aVDW*bVDW)/P
     #define d(V) 3*V*V-2*SE1*V+aVDW/P
 8
 9
      // REDLICH-KWONG EOS
10
     #define g(V) V*V*V-(R*T/P)*V*V-SE2*V-SE3
11
     #define h(V) 3*V*V-2*(R*T/P)*V-SE2
     // SOAVE REDLICH-KWONG EOS
12
13
     #define m(V) V*V*V-(R*T/P)*V*V-SE7*V-SE8
     #define n(V) 3*V*V-2*(R*T/P)*V-SE7
14
15
      // PENG ROBINSON EOS
16
      #define x(V) V*V*V+SE9*V*V-SE13*V+SE14
17
     #define y(V) 3*V*V+2*SE9*V-SE13
      int main()
18
19 🖂 {
20
          int option, step=1, N;
21
          float R, Tc, Pc, T, P, V_ideal, aVDW, bVDW, SE1, aREDL, bREDL, SE2, SE3;
22
          float aSRK, bSRK, Tr, w, SE4, SE5, SE6, alphaSRK, SE7, SE8;
23
          float aPENG, bPENG, SE9, SE10, SE11, SE12, alphaPENG, SE13, SE14;
24
          float V1, f0, f1, d0, e, g0, g1, h0, m0, m1, n0, x0, x1, y0;
25
26
          cout<<"\n"<<"COMPUTING VOLUME FROM THESE CUBIC EQUATION OF STATE"<<"\n"<<endl;</pre>
27
          cout<<setw(13)<<"(1) "<<"Van der Waals"<<endl;</pre>
          cout<<setw(13)<<"(2) "<<"Redlich-Kwong"<<endl;</pre>
28
          cout<<setw(13)<<"(3) "<<"Soave Redlich-Kwong"<<endl;</pre>
29
30
          cout<<setw(13)<<"(4) "<<"Peng Robinson"<<endl<<endl;</pre>
31
          cout<<"Choose Option: ";</pre>
32
          cin>>option;
          switch(option)
33
```

Fig. 1 Dev C++ Syntax showing various headers

Figure 1 defines the EOS and their derivatives (lines 6-17), as well as declaring many parameters as floats and int (lines 20-24). Equations 1-8 are defined in lines 6-17. SE1 to SE14 are expressions derived out of the general function,  $f(\hat{V})$ . It is to further reduce the complexity of the equation.

In C++ programming, 'int' stands for integer. Variables of type 'int' declared in line 20 of Figure 1 are examples. 'Float' is a decimal point number. Variables declared as 'float' are expected to be to some certain decimal places.

Output statements in C++ always begin with the keyword 'cout'. Lines 27-30 are output statements showing selectable options to execute. When the program is run, it requests the user to select an option, either 1, 2, 3, or 4, for any of the EOS (based on the input statement keyword 'cin'

of line 32). The user is expected to press either 1, 2, 3 or 4 to pick an equation of state. The options are well-interpreted using different cases in C++. They are Case 1 (line 35; Fig. 2), Case 2 (line 86; Fig. 2), Case 3 (line 138; Fig. 4a) and Case 4 (line 197; Fig. 4b).

C++ is case-sensitive. If a variable is declared using a lowercase alphabet, it should be declared so throughout the coding. Mistakes like that will prevent the program from running. This information is for readers who wish to type and run the source code in this article, either for usage or confirmation. It is worthy of note that almost all line of code in C++ ends with a semi-colon (;). Forgetting to type this will result in an error notification when the program is eventually run.

```
34 🗄
          ł
35
              case 1:
                  cout<<"\n\t"<<"Enter the following parameters:"<<endl<<endl;</pre>
36
37
                  cout<<setw(20)<<"T = ";</pre>
                  cin>>T;
38
39
                  cout<<setw(20)<<"P = ";
40
                  cin>>P;
41
                  cout<<setw(20)<<"Tc = ";</pre>
42
                  cin>>Tc;
43
                  cout<<setw(20)<<"Pc = ";</pre>
44
                  cin>>Pc;
45
                  cout<<setw(20)<<"R = ";</pre>
46
                  cin>>R;
47
                  aVDW=(27*pow(R,2)*pow(Tc,2))/(64*Pc);
                  bVDW=(R*Tc)/(8*Pc);
48
49
                  SE1=bVDW+(R*T)/P;
                  50
                  cout<<"NEWTON-RAPHSON METHOD";
51
                  52
53
                  cout<<setprecision(4)<<fixed;</pre>
54
               * INPUTS *
                  cout<<"\t"<<"Guess value, V_ideal = ";</pre>
55
56
                  cin>>V ideal;
                  cout<<"\t"<<"Specify Tolerable Error: ";</pre>
57
58
                  cin>>e;
59
                  cout<<"\t"<<"Maximum Iteration = ";</pre>
60
                  cin>>N;
                  cout<<"\n"<<setw(5)<<"Iteration"<<setw(7)<<"V"<<setw(10)<<"f(V)"<<setw(11)<<"f'(V)";</pre>
61
                      do
62
63 ⊟
                      Ł
                          f0=f(V_ideal);
                                              // = f(V)
64
                                              // = f'(V)
65
                          d0=d(V_ideal);
                          if(d0==0.0)
66
67 白
                           ł
68
                               cout<<"Mathematical Error";</pre>
69
                               exit(0);
70
71
                          V1=V ideal-f0/d0;
                                                   // Newton's Formula: xn+1=xn-f(xn)/f'(xn)
                           cout<<"\n"<<setw(5)<<step<<setw(13)<<V1<<setw(9)<<f(V1)<<setw(11)<<d(V1);
72
                                                    // Iteration incremented by '1
73
                           step=step+1;
74
                           V ideal=V1;
75
                          if(step>N)
76 🗖
                               cout<<"Not Convergent";
77
78
                               exit(0);
79
                           f1=f(V1);
80
81
82
                      while(fabs(f1)>e);
83
                      cout<<endl<<"Root is V = "<<V1<<endl;</pre>
84
                  break:
85
86
              case 2:
                  cout<<"\n\t"<<"Enter the following parameters:"<<endl<<endl;</pre>
87
                  cout<<setw(20)<<"T = ";
88
89
                  cin>>T;
90
                  cout<<setw(20)<<"P = ";
                  cin>>P;
91
92
                  cout<<setw(20)<<"Tc = ";</pre>
93
                  cin>>Tc;
94
                  cout<<setw(20)<<"Pc = ";</pre>
95
                  cin>>Pc;
96
                  cout<<setw(20)<<"R = ";
97
                  cin>>R;
```

#### Fig. 2 C++ syntax for cases 1 and 2

Figure 2 contains coding of the Van der Waal and Redlich-Kwong EOS as Case 1 and 2, respectively. Line 62-82 is a do-while loop for Newton's iteration. The user is allowed to specify the error, **e**, as well as the number of iterations or steps, N.



Fig. 3	Svntax	for	Redlich-	Kwong	execution

134

135

136

137

138

139

140

141 142

143

144

145

146

147

148

149

150

151

152

153

154

155

156

157

158

159

160 161

162

163 164

165

166

Abbreviations like aVDW, bVDW, simply imply 'a' and 'b' constant parameters for Van der Waals EOS. Similarly, that of Redlich-Kwong are subscripted as REDL, SRK for Soave-Redlich-Kwong and PENG for Peng-Robinson (See Figure 3, lines 98-101). Also, see Figure 1 (lines 21-23). The function in the form of 'pow(x,y)', as seen in lines 47, 98, 100-101, 152, 155-156, 211, 214, 216 and 218-220 in almost all the figures, defines variables, say' x' to the power of 'y'.

The program requests T, P, critical temperature and pressure, T<sub>c</sub>, and P<sub>c</sub>, molar gas constant, R, as well as an acentric factor,  $\omega$ , to be entered for program execution (one of such is line 139-151 of Figure 4a). It is left for the user to ensure unit consistency of the properties requested as well as their accuracy. A wrong entry will definitely give a wrong output.

Note that the coding is almost the same for all cases. Differences are in the written equations and parameter expressions (for instance, aSRK in line 152 differs from aPENG in line 211 of Figure 5).

```
133
                      while(fabs(g1)>e);
                      cout<<endl<<"Root is V = "<<V1<<endl;</pre>
                  break;
              case 3:
                  cout<<"\n\t"<<"Enter the following parameters:"<<endl<<endl;</pre>
                  cout<<setw(20)<<"T = ";</pre>
                  cin>>T:
                  cout<<setw(20)<<"P = ";</pre>
                  cin>>P;
                  cout<<setw(20)<<"Tc = ";</pre>
                  cin>>Tc;
                  cout<<setw(20)<<"Pc = ";</pre>
                  cin>>Pc;
                  cout<<setw(20)<<"R = ";</pre>
                  cin>>R;
                  cout<<setw(20)<<"w = ";
                  cin>>w;
                  aSRK=(0.42747*pow(R,2)*pow(Tc,2))/Pc;
                  bSRK=(0.08664*R*Tc)/Pc;
                  Tr=T/Tc:
                  SE4=0.48508+1.55171*w-0.15613*pow(w,2);
                  SE5=1-pow(Tr,0.5);
                  SE6=1+SE4*SE5;
                  alphaSRK=pow(SE6,2);
                  SE7=pow(bSRK,2)+(R*T*bSRK)/P-(aSRK*alphaSRK/P);
                  cout<<setprecision(4)<<fixed;</pre>
                 TNPLITS
                 cout<<"\t"<<"Guess value, V ideal = ";</pre>
```

The lengthy equations were shorten to a 'short expression' tagged 'SE', see lines 155-157 of Figure 4(a), so as to avoid mistakes in the equation entry that would clearly

affect the result accuracy. The precision is set to 4 (i.e. 4 decimal places) in line 164, to be maintained throughout the calculations.

167	cin>>V_ideal;					
168	<pre>cout&lt;&lt;"\t"&lt;&lt;"Specify Tolerable Error: ";</pre>					
169	cin>>e;					
170	<pre>cout&lt;&lt;"\t"&lt;&lt;"Maximum Iteration = ";</pre>					
171	cin>>N;					
172	cout<<"\n"< <setw(5)<<"iteration"<<setw(7)<<"v"<<setw(10)<<"f(v)"<<setw(11)<<"f'(v)";< td=""></setw(5)<<"iteration"<<setw(7)<<"v"<<setw(10)<<"f(v)"<<setw(11)<<"f'(v)";<>					
173	do					
174 📥	{{					
175	<pre>m0=m(V_ideal); // = f(V)</pre>					
176	n0=n(V ideal); // = $f'(V)$					
177	if(n0==0.0)					
178						
179	<pre>cout&lt;&lt;"Mathematical Error";</pre>					
180	exit(0);					
181 -						
182	V1=V ideal-m0/n0; // Newton's Formula: xn+1=xn-f(xn)/f'(xn)					
183	<pre>cout&lt;&lt;"\n"&lt;<setw(5)<<step<<setw(13)<<v1<<setw(9)<<m(v1)<<setw(11)<<n(v1);< pre=""></setw(5)<<step<<setw(13)<<v1<<setw(9)<<m(v1)<<setw(11)<<n(v1);<></pre>					
184	<pre>step=step+1; // Iteration incremented by '1'</pre>					
185	V ideal=V1;					
186	if(step>N)					
187 🗖	s scova)					
188	<pre>cout&lt;&lt;"Not Convergent";</pre>					
189	exit(0);					
190 -						
190	∫ m1=m(V1);					
191	m1=m(V1);					
192	<pre>while(fabs(m1)&gt;e);</pre>					
195	cout< <endl<<endl<<endl<<endl<<endl<< td=""></endl<<endl<<endl<<endl<<endl<<>					
194						
	break;					
196	4.					
197	case 4:					
198	<pre>cout&lt;&lt;"\n\t"&lt;&lt;"Enter the following parameters:"&lt;<endl<<endl;< pre=""></endl<<endl;<></pre>					
199	cout< <setw(20)<<"t ";<="" =="" td=""></setw(20)<<"t>					
200	cin>>T;					
201	cout< <setw(20)<<"p ";<br="" =="">Fig. (b) Continuation of Club Syntax for SDK Exception (Core 2)</setw(20)<<"p>					
	Fig. 4(b) Continuation of C++ Syntax for SRK Execution (Case 3)					

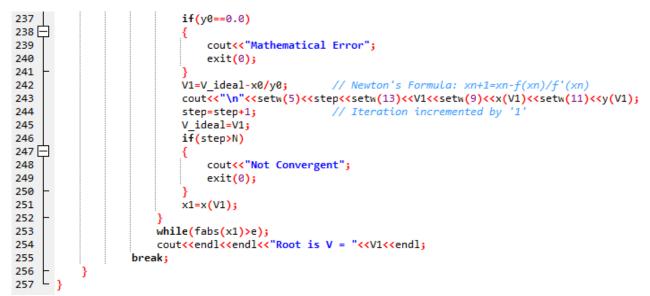
Figure 4b is a continuation of the codes for SRK EOS. The variable' m' and 'n' stands for  $f(\hat{V})$  and  $f'(\hat{V})$  corresponding to equations (5) and (6), respectively. The Newton expression in terms of these functions can be seen clearly in line 182. Initial guess was taken as  $V_{ideal}$  for the functions. Figure 5 has the execution for Peng Robinson EOS:



Fig. 5 C++ syntax for Peng-Robinson (case 4)

In line 215 of Figure 5, the reduced pressure,  $T_r$  has been accounted for. The user would not be asked to enter the value

of  $T_r$  but rather the critical temperature,  $T_c$  will be requested as  $T_r = \frac{T}{T}$ . Figure 6 is the remaining the coding:





The keyword break (line 255 of Figure 6) causes the program control to exit from the switch block. The keyword break must be included at the end of each case statement.

## **5. Program Execution**

## 5.1. Sample Problem

For propane, C<sub>3</sub>H<sub>8</sub>:  $T_c = 205.92^{\circ}\text{F} = 665.92R$ ,  $P_c = 615.5 \text{ psi}$  and acentric factor,  $\omega = 0.1529$  (Mansour), with number of moles, n = 1.136 lbmol,  $R = 10.73 \frac{\text{psi } ft^3}{\text{lbmol } R}$ , at temperature, T = 683R and pressure, P = 679.7 psi,

calculate the molar volume using the four CEOS. Do this for iteration number N = 10 and error, e = 0.001. Compare the result.

What, then, is the initial guess? The initial guess, as stated earlier, is,  $\hat{V} = \frac{RT}{P} = \frac{10.73 \times 683}{679.7} = 10.782 f^3/mol.$  This, together with five other variables, will be requested by the program when run. An additional variable (i.e. acentric factor,  $\omega$ ) will be requested by the program when the user chooses SRK and PENG equations of state.

	1. V	AN DER WA	AALS		2. R	EDLICH_KV	VONG
Choose Option: 1			Choose Opt	ion: 2			
Enter the following parameters:			Enter the following parameters:			arameters:	
	T =	683			T :	= 683	
	P =	679.7			P =	= 679.7	
	Tc =	665.92			Tc =	= 665.92	
	Pc =	615.5				= 615.5	
	R =	10.73			R =	= 10.73	
******	*******	*****		******	******	******	
NEWTON-RAPHSON METHOD			NEWTON-RAPHSON METHOD				
			10 792	*****	******	*****	
Guess value, V_ideal = 10.782		Gu	Guess value, V_ideal = 10.782				
Specify Tolerable Error: 0.001 Maximum Iteration = 10		Sp	Specify Tolerable Error: 0.001				
PI	aximum ite	Pation = 1	.0	Ма	ximum Ite	eration = 1	.0
		f(V)		Iteration	V	f(V)	f'(V)
		93.0404				111.2589	
2		27.9755		2		32.7285	
3		8.5002		3		9.3609	
4		2.5796				2.4075	
		0.7048		5		0.4306	
<u> </u>		0.1077				0.0262	
		0.0027		7		0.0202	
8	4.3058	0.0000	1.7574		4.0501	0.0001	4.2510
Root is V	= 4.3058			Root is V	= 4.6501		

	3. SOAVE-REDLICH-KW	ONG		4. PENG-	ROBINSON		
Choose Option: 3			Choose Option: 4				
En	ter the following p	parameters:	Ent	ter the f	ollowing p	arameters:	
				_			
	T = 683				683		
	P = 679.9				679.7		
	Tc = 665.92 Pc = 615.5				665.92		
	R = 10.73				: 615.5 : 10.73		
	W = 0.1529				0.1529		
	w = 0.1525			w -	0.1529		
******	*****		******	*******	*****		
NEWTON-RAP	NEWTON-RAPHSON METHOD			NEWTON-RAPHSON METHOD			
******	***********			***********			
	Guess value, V_ideal = 10.782			Guess value, V_ideal = 10.782			
Specify Tolerable Error: 0.001			Specify Tolerable Error: 0.001				
Ma	ximum Iteration = 1	10	Max	ximum Ite	eration = 1	0	
T++-	5/11	51.00					
		f'(V)	Iteration				
1 2	8.3851 110.1346 6.8008 32.1103				123.8681		
3	5.7810 8.9460				35.9876		
4	5.1840 2.1273		4		9.9288		
5	4.9250 0.3027				2.2958 0.3018		
	4.8741 0.0102				0.0084		
7	4.8722 0.0000		7		0.0000		
				11/040	0.0000	0.4002	
Root is V	= 4.8722		Root is V =	= 4.7045			

Table 2 is an amazing display of the program source code output window for each EOS selected. The window has a black background, and the contents are white. The molar volumes obtained are to four decimal places considering N =10. This program aims to be a CEOS calculator to be applied by Chemistry and Chemical Engineering students of Polytechnics and Universities. The results of Table 2 can always be tested manually using a calculator. At this point, there is a need to define some nomenclature:

## Nomenclature

Т	Temperature (absolute)
Р	Pressure
a, b	Constant parameter
$T_c$ , $P_c$	Critical temperature and pressure
R	Molar gas constant
$T_r$	Reduced temperature
k, α	Constant parameters
W	Acentric factor
n	No. of moles

#### 

Volume of fluid

# 6. Conclusion

V

The program was written using Dev-Cpp 5.11 TDM-GCC 4.9.2 version. The results are molar volume computations only by Newton's Iterative Method. They exclude temperature, pressure and number of moles determination. They also exclude alternative numerical techniques for such problems. For this program, entering the right number of iterations and percentage error determines the convergent of the initial guess to the desired root.

Users should note that the source code converges for either large or small error values depending on the number of iterations entered. Entering 10 as a number of iterations will not necessarily mean that C++ should display the result up to the  $10^{th}$  iteration. In general, this coding is for CEOS; it excludes other classes of EOS. Similar codes that incorporate other EOS by either C++ or other programming languages can be proposed for further studies.

This work is not discouraging manual calculations. It is, however, simplifying the rigorous calculation steps involved before arriving at a solution. It will serve as a confirmation calculator for students and teachers alike who might encounter CEOS problems in the course of their studies.

# Acknowledgement

Avalanche of gratitude goes to Mr. David Asimiakhuini, who mentored us on the basis of C++ Computer Programming at University of Maiduguri in the year 2014-2016. I specifically acknowledge Dr. Abdu Zubairu of the Department of Chemical Engineering, University of Maiduguri, for his lecture on Equations of State in the year 2014 when I was an undergraduate student of the institution. Your immense contributions are worthy of recognition. I throw more weight of recognition to M. Tech. Ahmad Abubakar Mustapha (a friend at VFSTR University, India) is also a C++ expert. Ultimately, glory be to Allah Almighty for granting us the wisdom to put this knowledge into use.

### References

- [1] Ulrich K. Deiters, and Ricardo Macías-Salinas, "Calculation of Densities from Cubic Equations of State: Revisited," *Industrial & Engineering Chemistry Research*, vol. 53, no. 6, pp. 2529–2536, 2014. [CrossRef] [Google Scholar] [Publisher Link]
- [2] Pallab Ghosh, Numerical Methods with Computer Programs in C++, PHI Learning Private Limited, 2009. [Google Scholar]
- [3] David Mautner Himmelblau, and James B. Riggs, *Basic Principles and Calculations in Chemical Engineering*, New Jersey: Prentice Hall, 1982. [Google Scholar]
- [4] Zakia Nasri, and Housam Binous, "Applications of the Soave-Redlich-Kwong Equation of State Using Mathematica," *Journal of Chemical Engineering of Japan*, vol. 40, no. 6, pp. 534–538, 2007. [CrossRef] [Google Scholar] [Publisher Link]
- [5] Zakia Nasri, and Housam Binous, "Applications of the Peng-Robinson Equation of State using MATLAB," *Chemical Engineering Education*, vol. 43, no. 2, pp. 115-124, 2009. [Google Scholar] [Publisher Link]
- [6] Mohamed Eman Mansour, Equation of State, Inverse Heat Conduction and Heat Exchangers, Cairo: Intech Open, 2020. [Google Scholar]
- [7] Attila Mate, *Introduction to Numerical Analysis with C Program*, Brooklyn College of the City University of New York, 2014. [Google Scholar] [Publisher Link]
- [8] M.B. Oliveira et al., "Modeling Phase Equilibria Relevant to Biodiesel Production: A Comparison of gE Models, Cubic EoS, EoS-gE and Association EoS," *Industrial & Engineering Chemistry and Research*, vol. 50, no. 4, pp. 2348-2358, 2011. [CrossRef] [Google Scholar] [Publisher Link]
- [9] Sashay Ramdharee, Edison Muzenda, and Mohamed Belaid, "A Review of the Equations of State and their Applicability in Phase Equilibrium Modeling," *International Conference on Chemical and Environmental Engineering*, pp. 84-87, 2013. [Google Scholar] [Publisher Link]
- [10] S H. Ratnawati, "Improvement of the Redlich-Kwong Equation of State by Modification of Co-Volume Parameter," *Reaktor*, vol. 13, no. 1, pp. 58-65, 2010. [Google Scholar] [Publisher Link]